

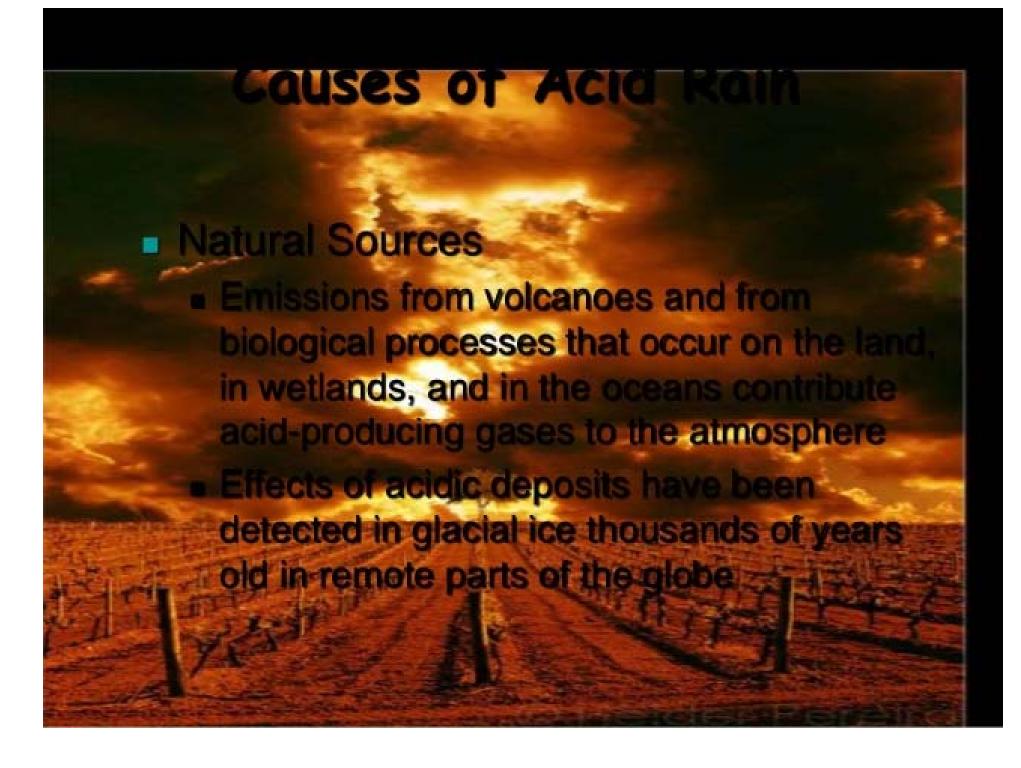
DEPARTMENT OF GEOGRAPHY CH. CHARAN SINGH UNIVERSITY, MEERUT

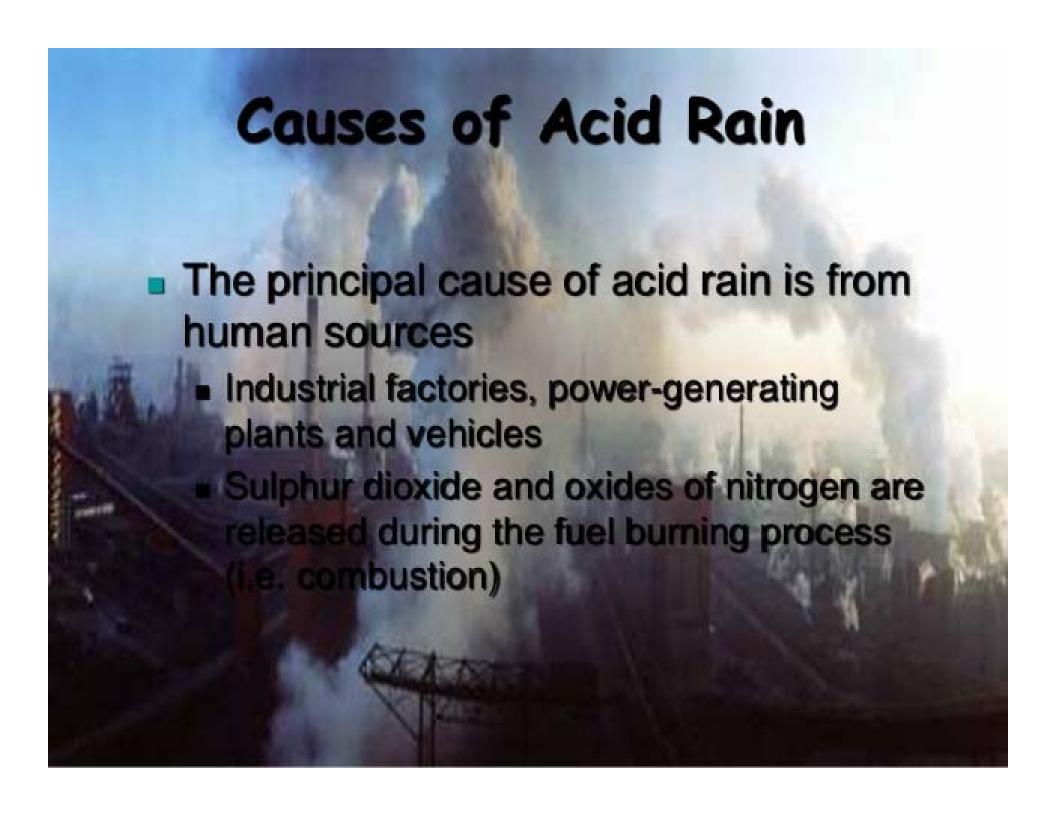
BY:

DR. SHALU
FACULTY
DEPARTMENT OF GEOGRAPHY

Definition of Acid Rain

- Precipitation that has a pH of less than that of natural rainwater (which is about 5.6 due to dissolved carbon dioxide).
- It is formed when sulphur dioxides and nitrogen oxides, as gases or fine particles in the atmosphere, combine with water vapour and precipitate as sulphuric acid or nitric acid in rain, snow, or fog.





Formation of Acid Rain

- When water vapour condenses, or as the rain falls, they dissolve in the water to form sulphuric acid (H₂SO₄) and nitric acid (HNO₃).
- While the air in cleaned of the pollutants in this way, it also causes precipitation to become acidic, forming acid rain







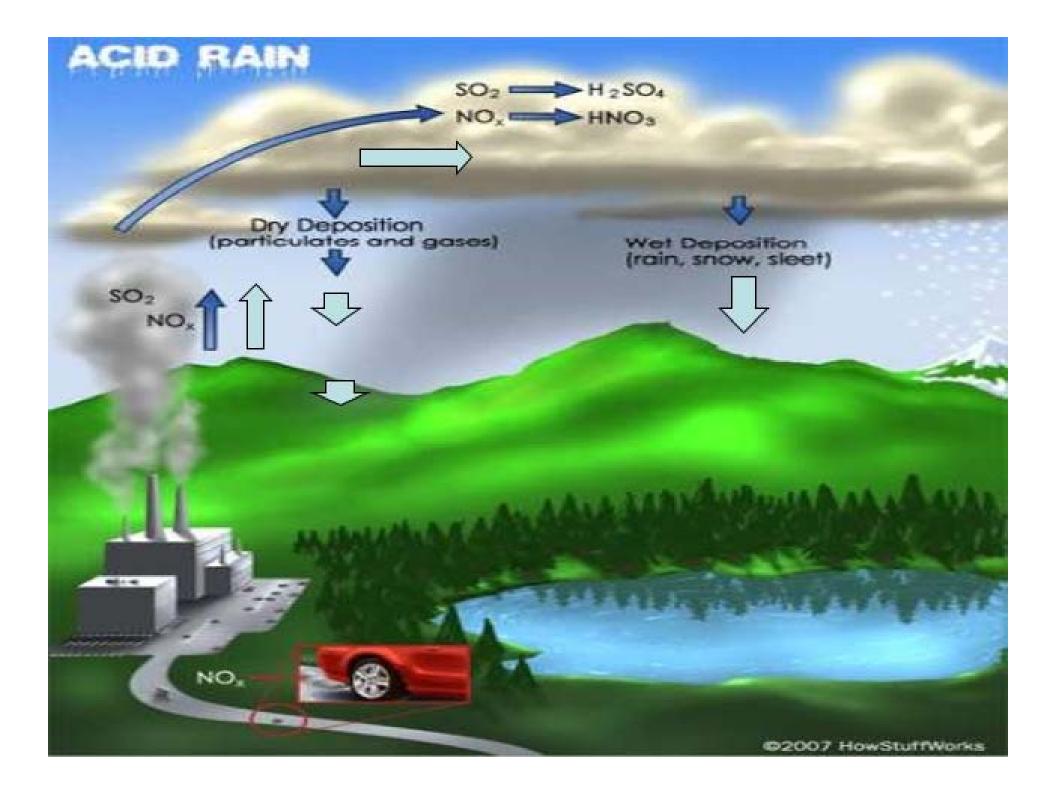




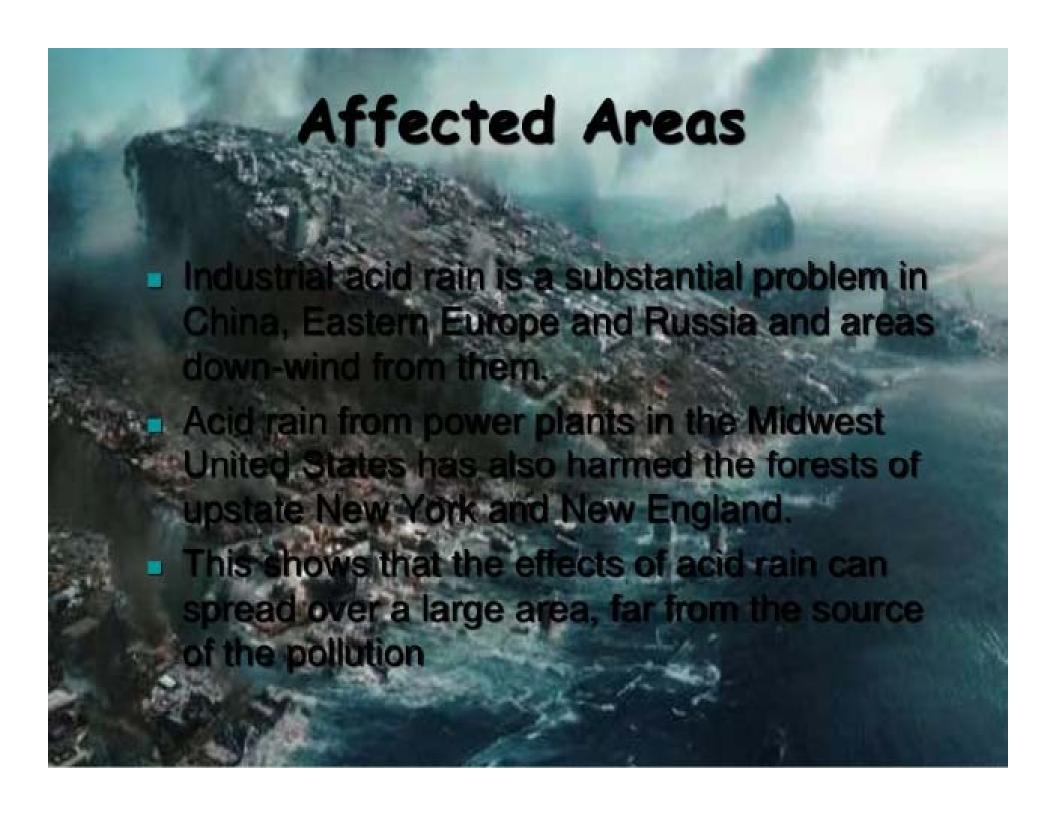


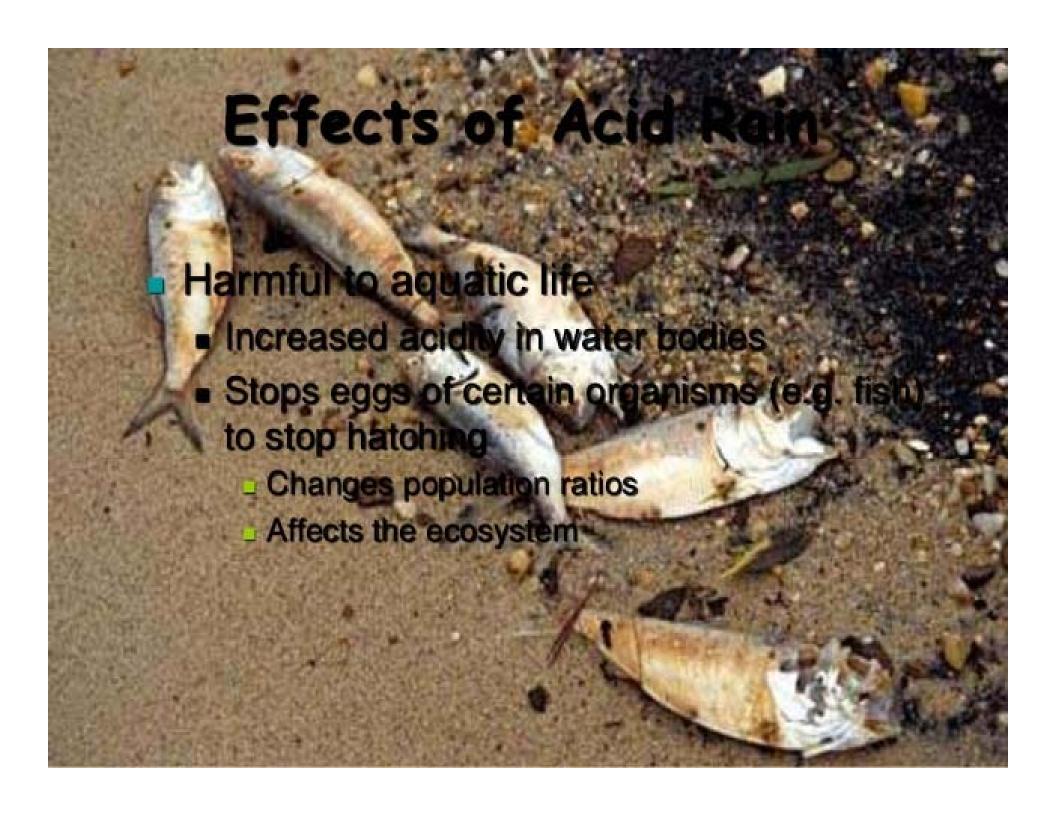


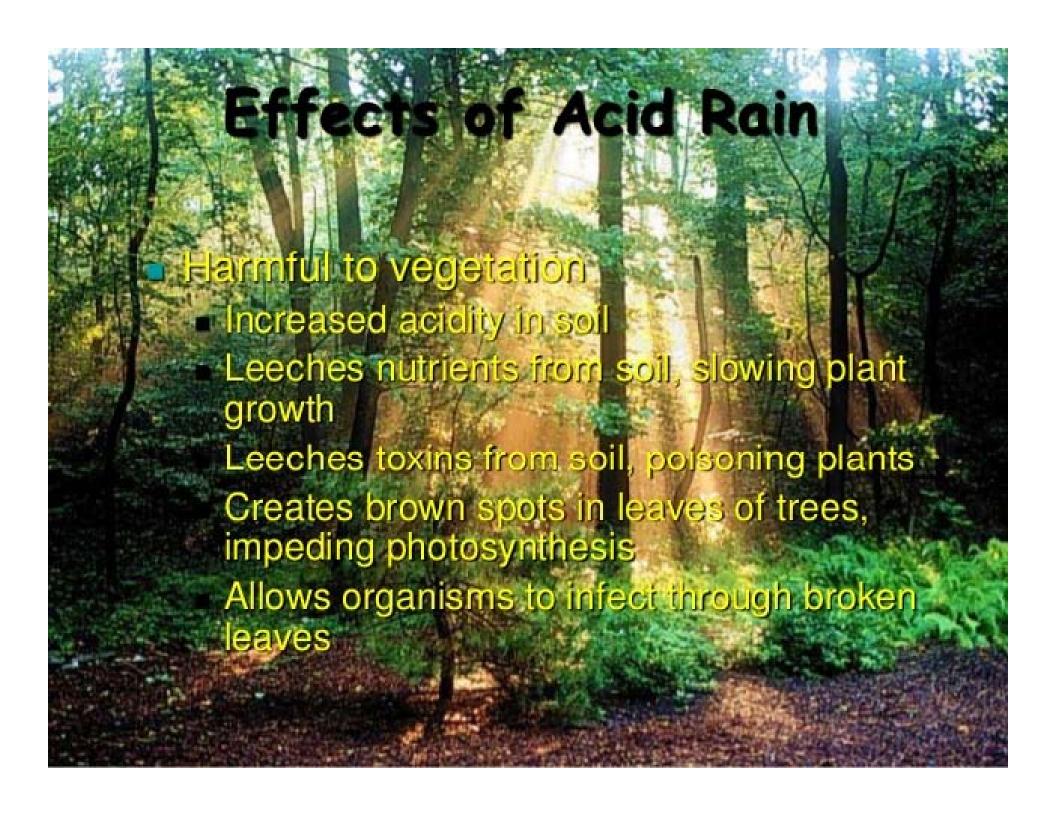












Effects of Acid Rain

- Accelerates weathering in metal and stone structures
 - Eg. Parthenon in Athens,
 Greece; Taj Mahal in Agra,
 India



















Effects of Acid Rain

Affects human health

- Respiratory problems, asthma, dry coughs, headaches and throat irritations
- Leeching of toxins from the soil by acid rain can be absorbed by plants and animals. When consumed, these toxins affect humans severely.
- Brain damage, kidney problems, and Alzheimer's disease has been linked to people eating "toxic" animals/plants.

















- Reduce amount of sulphur dioxide and oxides of nitrogen released into the atmosphere
 - Use less energy (hence less fuel burnt)
 - Use cleaner fuels
 - Remove oxides of sulphur and oxides of nitrogen before releasing
 - Flue gas desulphurization
 - Catalytic Converters

















- Flue Gas Desulphurization (FGD)
 - Removes sulphur dioxide from flue gas (waste gases)
 - Consists of a wet scrubber and a reaction tower equipped with a fan that extracts hot smoky stack gases from a power plant into the tower
 - Lime or limestone (calcium carbonate) in slurry form is injected into the tower to mix with the stack gases and reacts with the sulphur dioxide present









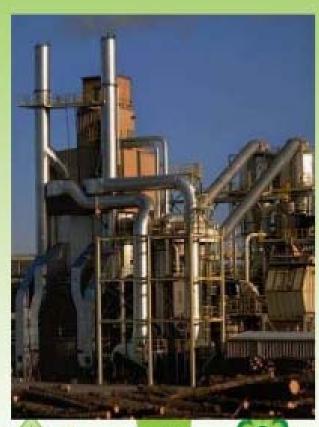








- (continued)
 - Produces pH-neutral calcium sulphate that is physically removed from the scrubber
 - Sulphates can be used for industrial purposes







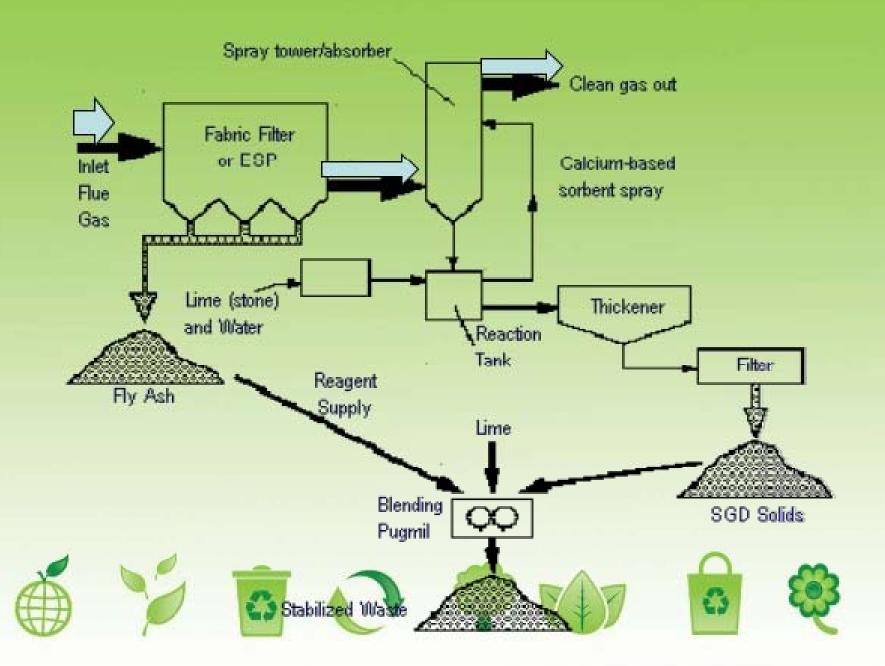








Scrubber at Work



 Use other sources of electricity (i.e. nuclear power, hydro-electricity, wind energy, geothermal energy, and solar energy)

Issue of cost













suggestion

The following are some more specific suggestion on what we, as an individual can do:

In the home

- •Run the dishwasher only with full load
- •Run the washing machine with full load
- •Buy energy-efficient appliances.
- Avoid the use of air conditioner altogether





suggestion

- Turn out the lights in empty rooms & when away from home
- Consider installing compact fluorescent bulbs instead of high-wattage incandescent bubs
- •Turn off the hot water tank when going away for extended period of time
- •Install additional insulation on hot water tank & pipes

















Suggestion

While shopping

Look for product bearing the EcoLogo. They minimize the use of environmentally hazardous substance & maximize energy efficiency & the use of recycled materials

















Suggestion

Transportation

- Have your vehicle engine tuned at least once every six months
- •Use alternative fuels, such as propane or natural gas
- Avoid unnecessary idling
- Drive at moderate speeds
- Take the train or bus on long trips
- Go CFC-free

















THANKING YOU FOR YOUR ACTICIPATION