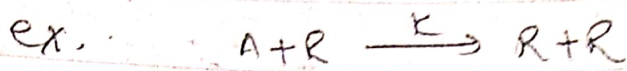


## # Autocatalytic Rxn

A reaction in which one of the product of rxn act as a catalyst is called an autocatalytic rxn.



$\therefore$  A formed R, and R remain constant in product (cat.)

$$-r_A = -\frac{dC_A}{dt} = k C_A C_R \quad \text{--- (1)}$$

Here, sum of moles remain constant.

$$C_{A0} + C_{R0} = C_A + C_R = C_0 \quad \text{--- (2)}$$

$\downarrow$   
constant

Note:  $C_{A0} \neq C_A$ ,  $C_{R0} \neq C_R$  (only sum will be same)

from eq (2)

$$C_R = C_0 - C_A$$

$$-r_A = k C_A (C_0 - C_A)$$

$$-\int_{C_{A0}}^{C_A} \frac{dC_A}{C_A (C_0 - C_A)} = k \int_0^t dt = kt$$

$$\Rightarrow \frac{1}{C_A (C_0 - C_A)} = \frac{A}{C_A} + \frac{B}{C_0 - C_A} \quad \Rightarrow 1 = A(C_0 - C_A) + B C_A$$

$$-\int_{C_{A0}}^{C_A} \left( \frac{1}{C_0} + \frac{1}{C_A} + \frac{1}{C_0} \cdot \frac{1}{C_0 - C_A} \right) dC_A = kt$$

$$-\frac{1}{C_0} \ln \frac{C_A}{C_{A0}} + \frac{1}{C_0} \left[ \ln(C_0 - C_A) \right]_{C_{A0}}^{C_A} = kt$$

$$-\frac{1}{C_0} \ln \frac{C_A}{C_{A0}} + \frac{1}{C_0} \ln \left[ \frac{C_0 - C_A}{C_0 - C_{A0}} \right] = kt$$

$$-\ln \frac{C_A}{C_{A0}} + \ln \left( \frac{C_0 - C_A}{C_0 - C_{A0}} \right) = C_0 kt$$

$$\ln \left[ \frac{(C_0 - C_A)}{C_0 - C_{A0}} \times \frac{C_{A0}}{C_A} \right] = C_0 kt$$

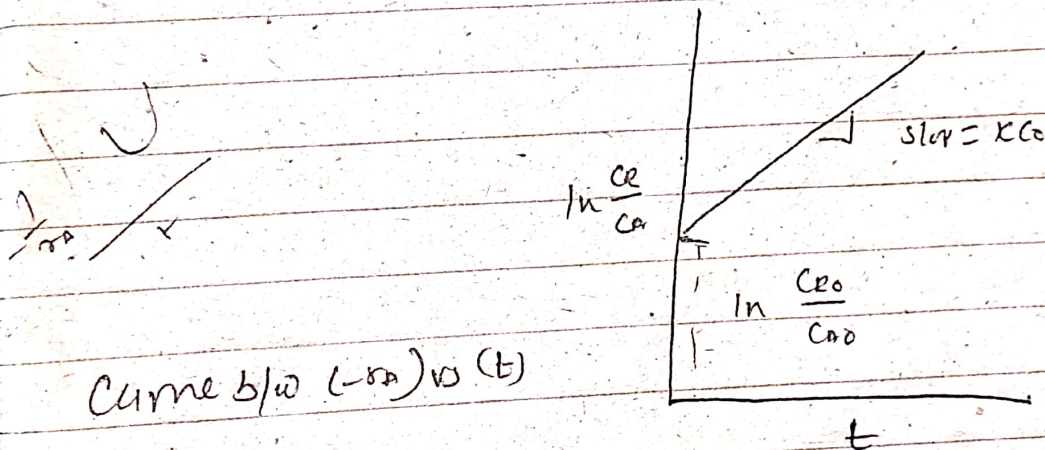
$$C_R + C_A = C_{R0} + C_{A0} = C_0$$

$$\ln \left[ \frac{C_R \cdot C_{A0}}{C_{R0} \cdot C_A} \right] = C_0 kt$$

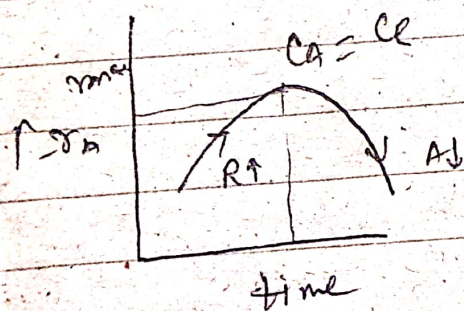
or

$$\ln \left( \frac{C_R}{C_A} \right) + \ln \frac{C_{A0}}{C_{R0}} = C_0 kt$$

$$\ln \frac{C_R}{C_A} = \ln \frac{C_{R0}}{C_{A0}} + C_0 kt$$



Curve b/w  $(-r_A)$  vs  $(t)$



$\Rightarrow$  In autocatalytic rxn product helps in increasing ROR

- i)  $C_R \uparrow$  then  $-r_A \uparrow$
- ii)  $C_A = C_R$  then  $-r_A = -r_{max}$
- iii) When  $C_A \downarrow$  then  $-r_A \downarrow$

## #1 Space Time & space velocity

- These term is used only flow reactors to measure their performance [PFR & MFR]

- space time ( $\tau$ ) - It is define as the time necessary to process one reactor volume of feed at specified condition.

- denoted by  $\tau$

- unit, - hr, minute, second

- Also define reciprocal of space velocity.

Example:- A space time of 5 minutes means that every 5 minutes one reactor volume of feed at specified condition is being treated by the reactor

### space velocity; (s)

It is define as the no. of the reactor volume of feed at specified condition which can be treated in unit time

- denoted by s.

-  $s = 1/\tau$  unit =  $\text{time}^{-1}$

Example:- A space velocity of  $2 \text{ min}^{-1}$  mean that two reactor volume of feed at specified condition one being fed into one reactor per minute.