

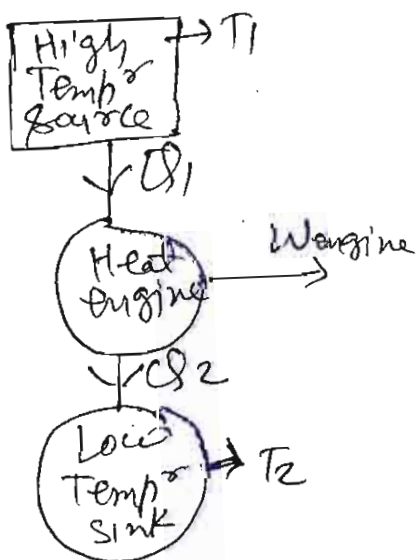
Agricultural Engg. IInd SEM, Thermodynamics, Refrigeration and Air Conditioning - by - Yogesh Kumar B-I

Topic \rightarrow Carnot's theorem \rightarrow It is impossible for any heat engine to be more efficient than a Carnot engine when operating between two given temperatures:-

Proof - $\eta_{\max} = \eta_{\text{Carnot}}$

Consider a heat engine drawing heat Q_1 from a heat reservoir at temperature T_1 , delivering work W and dumping (or ejecting) heat Q_2 into a heat sink at temperature T_2 . The heat engine operates in cycle, that is, it takes in heat Q_1 , does work W , rejects heat Q_2 , and in the end returns back to its original unchanged state.

Heat Engine and Efficiency -



Let $T_1 = \text{Temp}^{\circ}$ of the source (High)
 $T_2 = \text{Temp}^{\circ}$ of the sink (Low)
 $Q_1 = \text{Heat supplied by source to Heat engine}$
 $Q_2 = \text{Heat rejected to sink}$
 $W = \text{work output from Heat engine}$
From First Law of thermodynamics -

$\Rightarrow Q_1 - Q_2 = W$
Efficiency of engine, $\eta = \frac{\text{work output}}{\text{Heat supplied}}$

$$\Rightarrow \eta = \frac{W}{Q_1} = \frac{Q_1 - Q_2}{Q_1} = 1 - \frac{Q_2}{Q_1}$$

Consider, the net change in entropy ΔS of the universe

High Temp^r source releases a heat Q_1 at a constant Temp^r T_1 . Thus the change in its entropy is $\Delta S_1 = -\frac{Q_1}{T_1}$

because heat engine reject heat to the sink (Q_2) at constant Temp^r T_2 Thus the change in its entropy is

$$\Delta S_2 = \frac{Q_2}{T_2}$$

Thus the net change in entropy of the universe is $\Delta S = \Delta S_1 + \Delta S_2$

$$= \frac{Q_2}{T_2} - \frac{Q_1}{T_1}$$

using the second law of thermodynamics, $\Delta S \geq 0$, which implies

$$\frac{Q_2}{T_2} - \frac{Q_1}{T_1} \geq 0$$

$$\frac{Q_2}{T_2} \geq \frac{Q_1}{T_1}$$

$$\frac{Q_2}{Q_1} \geq \frac{T_2}{T_1}$$

$$1 - \frac{Q_2}{Q_1} \leq 1 - \frac{T_2}{T_1}$$

$$\eta \leq \eta_{Carnot} \Rightarrow \eta_{max} = \eta_{Carnot}$$

for any query - Contact 5/11 2-3 P.M

Since the left hand side represents the efficiency of the given heat engine (η) and right hand side is the efficiency of a Carnot engine,